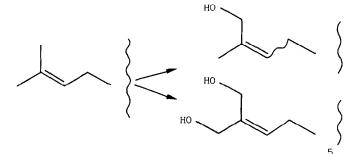
THE ACETATE OF (Z)-4-CHLORO-2-METHYL-2-BUTEN-1-OL STEREOSELECTIVE WITTIG SYNTHESIS OF A NEW HEMITERPENOID SYNTHON

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Summary a synthetic method for the preparation of the acetate of (Z)-4-chloro-2-methyl-2-buten-1-ol, a precursor of the (Z)-hydroxy prenyl synthon is described. The use of such synthon in the preparation of a metabolite of a prenyl-containing substance is also illustrated.

The hemiterpenoid group (isopentenyl, prenyl) present in various natural substances^{1,2} as well as in synthetic compounds of medicinal interest^{3,4} is well known to undergo oxidative attack during metabolic biotransformation in mammals



F and 7 - isomers

Recently, in our studies on the metabolism of feprazone $(\underline{1})^5$ an antiinflammatory drug containing a prenyl side-chain, three uninary metabolites in rat were isolated having the following structures

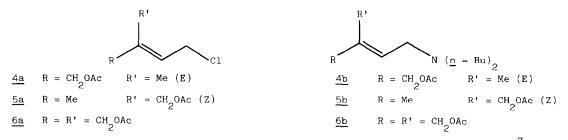


The synthetic strategy for a preparation of these substances ($\underline{2}$, $\underline{3}$, and $\underline{4}$) poses some challenges Without considering problems connected with stereoselectivity for $\underline{2}$ and $\underline{3}$, chemoselective oxidation⁶ of the terminal methyl groups of the prenyl part of the molecule is clearly

unpractical owing to the sensitivity of $\underline{1}$ to oxidizing agents On the other hand, the use of a Wittig reaction for double bond generation⁶ requires the synthesis of the suitable carbonyl compound of the rather complex 3,5-pyrazolidinedione molecule

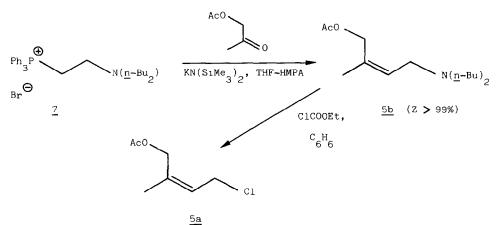
A synthon approach (represented by the bond dissection) was thought reasonable in view of the favourable nucleophilic character of the anion of the 3,5-pyrazolidinedione nucleus and of the foreseeable reactivity of the allylic substituent

The trisubstituted functionalized olefin synthons (4a-6a) were then envisaged suitable tools for our approach



Preparation of synthon $\underline{4a}$ of (E)-configuration was available from the literature⁷, while a procedure for the symmetric $\underline{6a}$ was recently described by us⁸ Their synthesis is based on the preparation of the pre-target allylic amines $\underline{4b}$ and $\underline{6b}$, respectively, from which the desired synthons are obtained by preferential splitting with ethyl chloroformate. Nothing is instead reported in the literature concerning the (Z)-synthon $\underline{5a}$ A correspondingly simple preparation of this synthon was therefore deemed of interest in view also of the potential utility of such intermediates in the field of substances containing the isoprene unit

In this letter, we wish to report the stereoselective synthesis of the acetate of (Z)-4---chloro-2-methyl-2-buten-1-ol (<u>5a</u>) <u>via</u> the corresponding key-intermediate <u>5b</u>, and illustrate its synthetic usefulness in the preparation of metabolite <u>2</u>



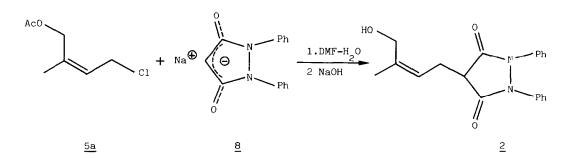
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When acetoxyacetone⁹ (0.21 mole) was allowed to react with the unstabilized ylide obtained from the β -dialkylamino phosphonium salt $\underline{7}^{10}$ (0.3 mole) and potassium hexamethyldisilazide (0.28 mole) under the conditions described by Sreekumar et al ¹¹ (10% HMPA/THF), the allylic acetate <u>5b</u> (51% yield, bp_{0.1} 85-87°) of high stereoisomeric purity was obtained <u>/</u>7isomer > 99% by vpc, DC 550 column, temp 170°C Retention time for <u>5b</u> (Z-isomer) 1227 sec in comparison with the corresponding <u>4b</u> (E-isomer)⁷ 1311 sec.<u>7</u> Splitting of the allylic substituent from <u>5b</u> by ethyl chloroformate (1.5 eq) in refluxing benzene for 2 hr gave, after distillation to effect separation from ethyl di-n-butyl-carbamate produced as a by-product, the stereoselectively pure Z-synthon <u>5a</u> <u>/</u>74% yield, bp₁₅ 98-101°, vpc analysis DC 550 column, temp 130°C Retention time 565 sec for <u>5a</u>, in comparison with the corresponding <u>4a</u> (E-isomer)⁷ 611 sec <u>7</u>

The structure of these hemiterpenoids follow from microanalysis, ir and nmr spectral data <u>5b</u> ir (film) 1740 cm⁻¹, nmr (CCl₄) δ 5 42 (t, 1H, CH₂ = C<u>H</u>), 4 53 (s, 2H, AcOC<u>H₂</u>), 3 02 (d, 2H, CH = C<u>H₂-N</u>), 2 33 (t, 4H, CH₂C<u>H₂N</u>), 2 (s, 3H, CH₃CO), 1 77 (s, 3H, C<u>H₃-C =</u>), 1 13-1 5 (m, 8H, CH₂-CH₂), 0 9 (t, 6H, C<u>H₃CH₂</u>)

ir (film) 1740 cm⁻¹, nmr (CDCl₃) δ 5 66 (t, 1H, CH₂ = C<u>H</u>), 4 66 (s, 2H, AcOC<u>H₂</u>), 4 15 (d, 2H, CH = C<u>H</u>₂Cl), 2.08 (s, 3H, CH₃CO), 1 83 (s, 3H, CH₃ - C =) Double bond geometry (Z) was confirmed, in addition to vpc information, by comparison of the nmr spectral data of <u>5a</u> with those of the corresponding E-isomer (<u>4a</u>)⁷ In particular, <u>5a</u> exhibited different relative shifts for the AcOC<u>H₂</u> and CH₃ - C = singlets, which are downfield with respect to those of the corresponding E-isomer (<u>4a</u>) δ 4.43 and 1.73, respectively

By utilizing the (Z)-synthon 5a, the febrazone metabolite 2 could be synthesized according to the following scheme



Reaction of <u>5a</u> with the sodium salt of 1,2-diphenyl-3,5-dioxopyrazolidine $(\underline{8})^{12}$ in aqueous dimethylformamide at room temperature for 12 hr gave the acetate of the desired <u>2</u> along with some unreacted <u>8</u> Direct saponification of this mixture with 5% sodium hydroxide (2 hr, room temperature) furnished after acidification and chromatography on silica-gel <u>/</u>dichloromethane--methanol (9 1, v/v)_7 the desired metabolite <u>2</u> as a colourless solid <u>/</u>31% overall yield, mp 141°_7, ir (nujol) 3400, 3200, 1750, 1720, 1590, 750, and 705 cm⁻¹, nmr (CDCl₃) δ 7 23 (s, 10H, arom protons), 5 27 (t, 1H, CH₂ = C<u>H</u>), 4.02 (s, 2H, C<u>H₂OH</u>), 3 48 (t, 1H, COCHCO), 2 92 (m, 2H, = CHC<u>H₂CH</u>), 1.73 (s, 3H, CH₂C), 1.56 (brs, 1H, OH)

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